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THE EFFECT OF PERMEANT IONS ON SINGLE CALCIUM CHANNEL ACTIVATION IN MOUSE NEUROBLASTOMA CELLS: ION-CHANNEL INTERACTION

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SUMMARY

- 1. Single low-threshold inactivating (LTI or T-type) Ca²⁺ channels of undifferentiated neuroblastoma cells (clone N1E-115) were investigated using the patch-clamp technique.
- 2. Single-channel conductance, g_i , for Ca²⁺, Sr²⁺ or Ba²⁺ as a permeant cation was similar (7·2 pS). Mean channel open time, $\tau_{\rm op}$, was also practically independent of the divalent ion species; it decreased from 0·7 to 0·3 ms between -40 and 0 mV.
- 3. Modification of the calcium channel selectivity by lowering the external $\mathrm{Ca^{2+}}$ concentration to 10^{-8} M produced an increase in g_i for $\mathrm{Na^+}$ and $\mathrm{Li^+}$ ions and a shift of potential-dependent characteristics in the hyperpolarizing direction. Voltage sensitivity and absolute values of τ_{op} were also changed. These changes were dependent on both permeant monovalent ion type and concentration.
- 4. At high [Na⁺]_o, $\tau_{\rm op}$ was almost potential independent ($\simeq 0.3$ ms). Decrease in [Na⁺]_o and substitution of Li⁺ for Na⁺ increased $\tau_{\rm op}$ and the steepness of its potential dependency.
- 5. The divalent and monovalent cations that were tested had much smaller effect on the mean intraburst shut time, $\tau_{\rm cl(f)}$, which was nearly independent of membrane potential ($\simeq 0.6$ ms). By contrast, mean burst duration was strongly potential dependent and noticeably affected by permeant ion type.
- 6. All kinetic changes were analysed in terms of a four-state sequential model for channel activation. According to this model the channel enters the open state through three closed states. Transitions between closed states can be formally related to the transmembrane movement of two charged gating particles (m^2 process). The interaction between ion flux and a sterical region of the Ca²⁺ channel selectivity filter may, depending on ion transfer rate and ionic radius, lead to a local increase of the dielectric constant, resulting in redistribution of the electric field and changes in potential dependency of $\tau_{\rm op}$.

INTRODUCTION

In previous investigations of Ca²⁺ channel selectivity and gating at the whole-cell level, it has been shown that the inactivation time course of current is strongly affected by permeant ion species and concentration (Tillotson, 1979; Ashcroft & Stanfield, 1981; Brown, Morimoto, Tsuda & Wilson, 1981; Eckert & Tillotson, 1981; Ganitkevich, Shuba & Smirnov, 1987). This effect, however, has most often been explained by the involvement of Ca²⁺-dependent inactivation mediated through the cellular metabolic pathways rather than by direct ion-channel interaction (e.g. Chad & Eckert, 1986). The whole-cell data concerning the effect of permeating ions on Ca²⁺ channel activation and deactivation ('tail' current) are contradictory. Some authors have reported that there is no obvious effect of permeant ions on Ca²⁺ channel activation-relaxation kinetics (Hagiwara, Ozawa & Sand, 1975; Byerly & Hagiwara, 1982; Fox, Nowycky & Tsien, 1987a; Carbone & Lux, 1988) while others have noted distinct changes (Fenwick, Marty & Neher, 1982; Siami & Kung, 1982; Brown, Tsuda & Wilson, 1983; McDonald, Cavalie, Trautwein & Pelzer, 1986; Kostyuk & Shirokov, 1989). Although this disagreement is likely, in part, to be due to tissue specificity of Ca²⁺ channels, it is our view that the whole-cell experiments do not allow adequate resolution of all kinetic stages of the activation process (Kostyuk, Shuba & Teslenko, 1989b). Indeed, in almost all studies carried out at the single Ca²⁺ channel level, an effect of permeant ions on mean open time has been detected (Cavalie, Ochi, Pelzer & Trautwein, 1983; Nelson, 1986; Carbone & Lux, 1987b). However, a detailed investigation is still required, especially one to relate the changes in Ca²⁺ channel gating to a definite kinetic model of its activation.

It is well established that the Ca²⁺ permeability of the neuronal membrane is due to the activity of two (Veselovsky & Fedulova, 1983; Carbone & Lux, 1984; Fedulova, Kostyuk & Veselovsky, 1985) or possibly three (Nowycky, Fox & Tsien, 1985; Fox et al. 1987 a, b; Kostyuk, Shuba & Savchenko, 1988 a) types of potentialdependent Ca²⁺ channels differing in their voltage dependence, kinetics, selectivity and pharmacology. We have found that undifferentiated neuroblastoma cells (clone N1E-115) possess Ca²⁺ channels which have properties very similar to low-threshold inactivating (LTI) Ca²⁺ channels of vertebrate sensory neurones. In the present investigation the properties of these unitary LTI Ca²⁺ channels were studied using the patch-clamp technique. The changes in channel gating due to substitution of different permeant divalent and monovalent ions were analysed in terms of a fourstate sequential model for channel activation (Kostyuk et al. 1989b). Monovalent ion permeability was induced by lowering the external Ca²⁺ concentration to 10⁻⁸ M (Kostyuk, Mironov & Shuba, 1983; Almers, McCleskey & Palade, 1984; Fukushima & Hagiwara, 1985; Hess, Lansman & Tsien, 1986; Carbone & Lux, 1987b). LTI Ca²⁺ channels are especially convenient for the study of ion-channel interaction because they do not reveal Ca2+-dependent inactivation and are independent from intracellular metabolic support (Carbone & Lux, 1984, 1987b; Quandt & Narahashi, 1984; Fedulova et al. 1985; Kostyuk et al. 1988a). This allows us to exclude any indirect action of permeant ions on channel gating.

Preliminary results have been presented (Kostyuk, Shuba, Savchenko & Teslenko, 1988b; Kostyuk, Akaike, Osipchuk, Savchenko & Shuba, 1989a).

METHODS

Preparation. Mouse neuroblastoma cells (clone N1E-115, line C 1300) were cultured on cover-slips placed in Petri dishes filled with DMEM cultured medium (Serva) supplied with glucose (4·5 g/l) and 10% of fetal calf serum.

Electrophysiology and solutions. Experiments were carried out using the patch-clamp technique (Hamill, Marty, Neher, Sakmann & Sigworth, 1981). Macroscopic currents were recorded in the whole-cell configuration. In this case the external solution contained either Ca²⁺, Sr²⁺ or Ba²⁺ as a charge carrier and had the following composition (mm): CaCl₂ (SrCl₂ or BaCl₂), 15; MgCl₂, 4; tetraethylammonium chloride (TEA-Cl), 110; HEPES, 10; tetrodotoxin (TTX), 10⁻³. The pH was adjusted to 7·3 by NaOH. The recording pipette was filled with intracellular saline containing (mm): CsCl, 60; caesium aspartate, 60; Cs-EGTA, 10; HEPES, 20; MgCl₂, 4; pH 7·2. Analog compensation of leakage current was used when recording whole-cell current.

Single-channel recording was performed mostly in the cell-attached configuration. The resting potential (V_r) was brought to zero by using an extracellular solution containing a high concentration of K⁺ (mm): potassium aspartate, 140; K-EGTA, 10; HEPES, 20; MgCl₂, 5; pH 7·3. During single Ca^{2+} channel recording either 60 mm- Ca^{2+} , Sr^{2+} or Ba^{2+} was presented in the pipette solution as permeant cation; 2 mm-MgCl₂, 40 mm-TEA-Cl, 10 mm-HEPES and 10^{-6} m-TTX were also added and the pH was titrated to 7·3 with NaOH. When recording the activity of single Ca^{2+} channels with modified selectivity, the free Ca^{2+} concentration in the pipette solution was buffered to 10^{-8} m. At this concentration Ca^{2+} ions have no effect on the function of modified single Ca^{2+} channels (Carbone & Lux, 1987b). Na⁺ and Li⁺ ions at various concentrations were used as charge carriers through the modified Ca^{2+} channels. The basic composition of the pipette solution was as follows (mm): NaCl, 60; TEA-Cl, 85; $CaCl_2$, 0·91; $CaCl_2$, 0·91;

Data analysis. Single-channel currents were low-pass filtered at 1 or 2 kHz and digitized at a frequency of 10 kHz. Capacitative and leakage currents were eliminated by subtracting from each record the average of records without channel activity. Half-amplitude threshold was used for detection of channel openings and closings. With such a threshold, the 'dead time', $t_{\rm d}$ (maximum pulse duration in which current amplitude does not reach the threshold), is related to the corresponding bandwidth, Δf , as $t_{\rm d}=0.183/\Delta f$ (Colquhoun & Sigworth, 1983). Standard deviation of background noise fluctuations, $\sigma_{\rm n}$, was about 0.05 pA at $\Delta f=1$ kHz increasing to $\simeq 0.08$ pA at $\Delta f=2$ kHz. Unitary current amplitude, A_0 , for divalent charge carriers decreased from 0.66 to 0.26 pA in the range of membrane potentials -50 to 0 mV. This means that the relative threshold, $A_0/2\sigma_{\rm n}$, is about 6 at a membrane potential $V_{\rm m}=-50$ mV and only 2.5 at $V_{\rm m}=0$ mV. Thus, the rate of detection of false events, $\lambda_{\rm f}$, increases with the depolarization in this potential range from about $10^{-4}~{\rm s}^{-1}$ to $50~{\rm s}^{-1}$ (Colquhoun & Sigworth, 1983). For the Na⁺ current through the modified Ca²⁺ channels at [Na⁺] $_0=40~{\rm mM}$ and $V_{\rm m}=-40~{\rm mV}$ (worst case), $A_0/2\sigma_{\rm n}$ is of the order of 2.5 and $\lambda_{\rm f}\simeq 50~{\rm s}^{-1}$. Distortions of measured time constants related to false openings of short duration were avoided by excluding the initial part (0.2 ms) of open time and burst duration histograms from the fitting procedure.

For the experimental conditions when the highest rate of false openings occurred ($\lambda_{\rm f} \simeq 50~{\rm s}^{-1}$) an additional closed period longer than 20 ms might contribute to the closed time and first latency histograms. However, it was not difficult to distinguish these periods from real closings since they were distributed randomly and their duration was 2–3 times longer than the 'very slow' ($\tau_{\rm cl(vs)}$) time constant of these histograms (see Table 1). So, by choosing of the appropriate part of closed time and first latency histograms for the fitting procedure it was possible to take into account all real channel closings and to neglect false ones.

Using half-amplitude threshold, original records were idealized. Idealized records were used for construction of averaged currents and time histograms. To identify bursts, all shut intervals which were equal to or less than three times the mean duration of the 'fastest' component of the closed time histogram ($\tau_{\rm cl(f)}$) were considered to be gaps within bursts (Magleby & Pallota, 1983).

I-V relations of unitary Ca²⁺ channel for monovalent permeant cations were constructed from responses of membrane patches stimulated with voltage ramps. Discrete open channel amplitudes obtained from ramp records were averaged and represented in I-Vco-ordinates after digital filtering to reduce noise.

RESULTS

Divalent charge carriers

With 15 mm-Ca²⁺, Sr²⁺ or Ba²⁺ in the external Na⁺, K⁺-free solution the depolarization of the neuroblastoma cell membrane from a holding potential $(V_{\rm h})$ of -90 mV evoked an inward-going whole-cell current that was activated at $V_{\rm m}=-60$ to -40 mV (Fig. 1A). With increasing depolarization, current amplitude increased

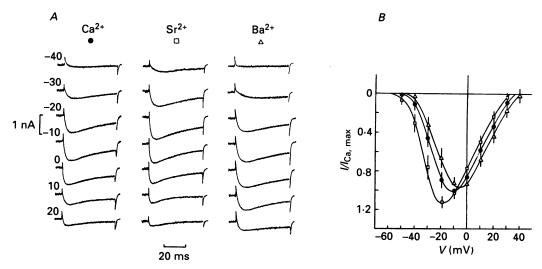


Fig. 1. Whole-cell currents through Ca^{2+} channels of neuroblastoma cells for divalent charge carriers, Ca^{2+} , Sr^{2+} and Ba^{2+} , all at 15 mm. A, original current records for indicated membrane potentials (mV). B, averaged I-V relations (mean \pm s.d. from seven cells normalized to the maximum Ca^{2+} current $I_{Ca, max}$).

and reached its maximum at $V_{\rm m}=-20$ to 0 mV. At higher depolarizations, current decreased again. The apparent reversal was observed at $V_{\rm m}=+30$ to +40 mV (Fig. 1B). Since the composition of the external and internal solutions ensured that other permeabilities were suppressed, these currents were attributed to the activation of Ca²⁺ channels. In contrast to dorsal root ganglion (DRG) neurons (Veselovsky & Fedulova, 1983; Carbone & Lux, 1984, 1987a; Fedulova et al. 1985) and differentiated neuroblastoma cells (Fishman & Spector, 1981; Narahashi, Tsunoo & Yoshii, 1987) only one component of inward current was observed under our conditions in the range of membrane potentials -90 to +80 mV. This suggests that only one type of potential-operated Ca²⁺ channel occurs in our preparation.

After the onset of depolarization, the current carried by Ca^{2+} , Sr^{2+} and Ba^{2+} ions reached a maximal value and then inactivated with time. The amplitudes of whole-cell currents for different divalent charge carriers were related to each other in the following way: $I_{Ca}:I_{Sr}:I_{Ba}=1:1\cdot12:0\cdot94$ (Fig. 1B). This sequence differs from that usually found for high-threshold Ca^{2+} channel currents in a large variety of preparations which has the highest amplitude in Ba^{2+} -containing solution and decreases after substitution of Sr^{2+} or Ca^{2+} for Ba^{2+} (Bean, 1985; Bossu, Feltz &

Thomann, 1985; Fedulova et al. 1985; Matteson & Armstrong, 1986; Carbone & Lux, 1987a; Fox et al. 1987a). On the other hand, a similar sequence to that determined here has been reported for whole-cell currents through LTI Ca²⁺ channels in lymphocytes (Fukushima & Hagiwara, 1985), vertebrate sensory neurons (Bossu et al. 1985; Carbone & Lux, 1987a, b), glomerulosa cells (Durroux, Gallo-Payet & Payet, 1988) and heart cells (Bean, 1985).

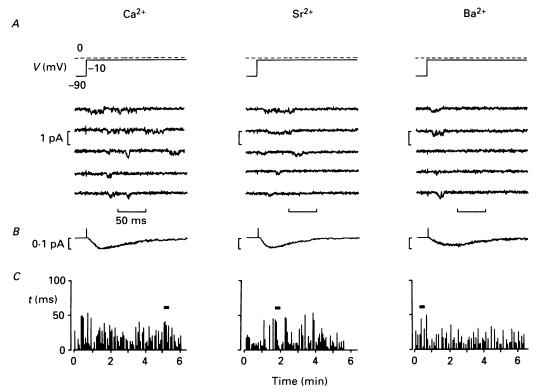


Fig. 2. Single $\operatorname{Ca^{2+}}$ channel activity in neuroblastoma cell membrane for divalent charge carriers. A, voltage protocols and original current records for $\operatorname{Ca^{2+}}$, $\operatorname{Sr^{2+}}$ and $\operatorname{Ba^{2+}}$ (60 mm) as charge carriers. Cell-attached configuration, resting membrane potential was zeroed. Recording bandwidth, 1 kHz. B, corresponding averages of computer-idealized current records. C, changes in total channel open time t in response to consecutive depolarization of 400 ms duration at 0·25 Hz (height of the columns) during the course of the experiment. Horizontal bars indicate records shown in A.

The I-V relation of the whole-cell current was shifted by permeant divalent cations along the voltage axis in the depolarizing direction (measured as a change in the potential of half-maximal current, $V_{\frac{1}{2}}$) according to the sequence $\mathrm{Ba^{2+}} > \mathrm{Ca^{2+}} > \mathrm{Sr^{2+}}$. This sequence is also quite similar to that reported for LTI current in other preparations ($\mathrm{Ba^{2+}} \simeq \mathrm{Ca^{2+}} > \mathrm{Sr^{2+}}$; Bean, 1985; Fukushima & Hagiwara, 1985), but different from that for high-threshold current ($\mathrm{Ca^{2+}} > \mathrm{Sr^{2+}} > \mathrm{Ba^{2+}}$; Fedulova et al. 1985). We also found a difference in apparent reversal potential of the corresponding currents. This was, however, almost the same as the $V_{\frac{1}{2}}$. That is why we would not

attribute shifts of reversal potential to different permeabilities for the divalent cations.

Figure 2A shows single Ca^{2+} channel currents for divalent charge carriers. Ensemble averages varied in amplitude as the corresponding whole-cell currents (Fig. 2B), although unitary current amplitudes in Ca^{2+} , Sr^{2+} and Ba^{2+} solutions were practically identical (Fig. 3D) as were channel slope conductances (7·2 pS). In

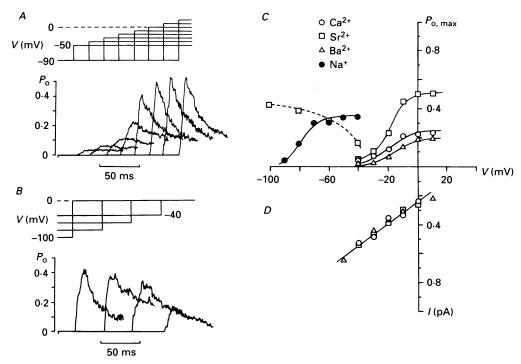


Fig. 3. Stationary characteristics of single Ca²+ channel of neuroblastoma cells. A, the probability of the channel being open, $P_{\rm o}$, as a function of time for various test potentials from holding potential $-90~{\rm mV}$ (60 mm-Sr²+). B, the same as A but in response to variation of holding potential at a fixed test potential (0 mV). Each $P_{\rm o}$ curve was generated using 80–120 single-channel records for corresponding potentials with subsequent averaging of the data from three to five patches. Voltage protocols are shown at the top of A and B. C, Ca²+ channel activation curves (maximal $P_{\rm o}$, $P_{\rm o,max}$ as a function of membrane potential) for Ca²+, Sr²+, Ba²+ and Na+ (60 mm). Dashed line indicates the curve of steady-state inactivation (60 mm-Sr²+). D, Ca²+ channel I-V relations for different divalent charge carriers (60 mm); conductance, 7·2 pS. Different symbols in C and D indicate mean experimental values for corresponding charge carriers; continuous lines are the best fit by eye.

addition, there was no significant change in the total channel open time t during successive depolarizations when one permeant ion was substituted for another (Fig. 2c). However, there were more blank records when Ba^{2+} was used as a charge carrier compared to Ca^{2+} or Sr^{2+} . As a result, averaged currents revealed variation in amplitude, and activation curves of the channel for Ba^{2+} , Ca^{2+} and Sr^{2+} had different saturation levels of 0.2, 0.3 and 0.5, respectively. The observed relative shifts of

channel activation curves by different permeant ions were in qualitative agreement with those for I-V relations of the corresponding whole-cell currents (Fig. 3C).

During single-channel recording with divalent charge carriers the $V_{\rm h}$ was set at -90 mV. Shifting $V_{\rm h}$ in the depolarizing direction caused stationary inactivation of the channel (Fig. 3B and C). Half-inactivation was observed at about -40 mV for ${\rm Sr}^{2+}$ as a permeant cation.

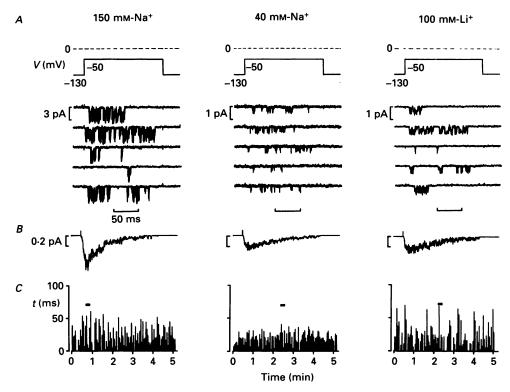


Fig. 4. Activity of modified Ca^{2+} channel of neuroblastoma cells. Na^{+} (150 and 40 mm) and Li^{+} (100 mm) as charge carriers. Recording bandwidth 2 (Na^{+}) and 1 (Li^{+}) kHz. Details as in Fig. 2.

Monovalent charge carriers

Figure 4 shows examples of single-channel activity in neuroblastoma cell membrane after a decrease in external $\mathrm{Ca^{2+}}$ concentration to 10^{-8} m. $\mathrm{Na^{+}}$ and $\mathrm{Li^{+}}$ were used as charge carriers. Since there are no TTX-resistant $\mathrm{Na^{+}}$ channels in neuroblastoma cells (Moolenaar & Spector, 1978; Fishman & Spector, 1981; Quandt & Narahashi, 1984; Narahashi *et al.* 1987), we attributed this activity to single $\mathrm{Ca^{2+}}$ channels with modified selectivity. The main features of modified $\mathrm{Ca^{2+}}$ channels were a strong increase in the conductance and a substantial shift of the activation characteristic in the hyperpolarizing direction (~ 70 mV, Fig. 3C). The I-V relations of modified $\mathrm{Ca^{2+}}$ channels obtained from ramp records for $\mathrm{Na^{+}}$ concentrations in the range of 40-150 mm and $\mathrm{Li^{+}}$ at 100 mm are presented in Fig. 5. Under our experimental conditions when total current through the modified $\mathrm{Ca^{2+}}$ channel

consisted of inward Na⁺ or Li⁺ and outward K⁺ currents, all I-V relations were linear up to $V_{\rm m}=+30$ mV, and channel conductances for 40, 60, 100 and 150 mm-Na⁺ and 100 mm-Li⁺ were 10, 18, 32 and 57 pS and 11 pS, respectively. In contrast to observations by Hess *et al.* (1986) on single high-threshold non-inactivating (HTN or

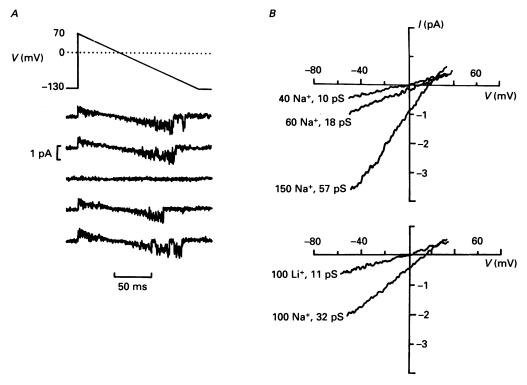


Fig. 5. Ramp recordings of modified Ca^{2+} channel. A, voltage protocol and original current records (60 mm-Na⁺ as a charge carrier). B, modified Ca^{2+} channel I-V relations obtained after averaging of discrete channel openings from ramp recordings and digital filtering of resultant curves to reduce noise (corresponding concentrations (mm) and conductance of charge carriers are shown near each curve).

L-type) $\mathrm{Ca^{2+}}$ channels of guinea-pig ventricular myocytes, we could detect clear unitary outward current at depolarizations to potentials more positive than the reversal potential. However, at V_{m} higher than $+40~\mathrm{mV}$ multiple unitary events appeared relatively frequently. They could arise from the presence of more than one $\mathrm{Ca^{2+}}$ channel in the patch since rare superimpositions were observed also at more negative V_{m} . A contribution of $\mathrm{K^{+}}$ channels to the unitary outward currents seems to be improbable since in most cases the pipette solution contained relatively high concentrations of TEA (except in $[\mathrm{Na^{+}}]_{\mathrm{o}} = 150~\mathrm{mm}$) and only less than 10% of all patches contained channel activity consisting of outward currents without accompanying inward currents. For the construction of I-V relations, we tried to choose only those ramp records which did not contain apparent overlapping events. Nevertheless, we are uncertain of current values at membrane potentials more positive than $+30~\mathrm{to}+40~\mathrm{mV}$.

The sequence of relative permeabilities for monovalent cations was obtained by using the following expressions derived from the Goldman-Hodgkin-Katz equation (Hille, 1975):

$$\begin{split} V_{\mathrm{Na}} &= RT/F \ln{(P_{\mathrm{Na}}[\mathrm{Na^+}]_{\mathrm{o}}/P_{\mathrm{K}}[\mathrm{K^+}]_{\mathrm{i}})}, \\ V_{\mathrm{Na}} - V_{\mathrm{Li}} &= RT/F \ln{(P_{\mathrm{Na}}[\mathrm{Na^+}]_{\mathrm{o}}/P_{\mathrm{Li}}[\mathrm{Li^+}]_{\mathrm{o}})}, \end{split}$$

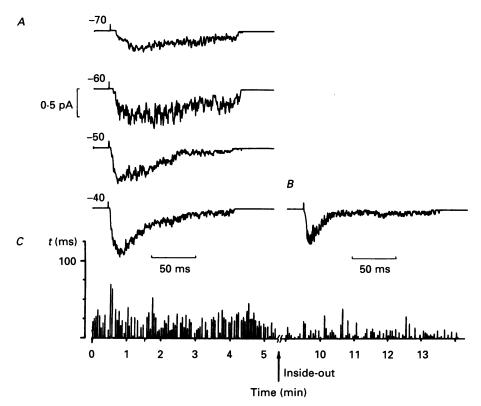


Fig. 6. Effect of patch excision on the activity of modified $\operatorname{Ca^{2+}}$ channel of neuroblastoma cells. A, averages of computer-idealized single-channel records for indicated membrane potentials (mV) obtained in cell-attached configuration (100 mm-Na⁺ as a charge carrier). B, the same for a membrane potential of -40 mV after formation of inside-out configuration. C, changes of total channel open time t in response to consecutive depolarizations during the course of the experiments.

where $V_{\rm Na}$, $V_{\rm Li}$ are reversal potentials for Na⁺ and Li⁺ currents, $P_{\rm Na}$, $P_{\rm K}$, $P_{\rm Li}$ are permeabilities, [Na⁺]_o and [Li⁺]_o are external and [K⁺]_i internal ion concentrations, and R, T and F have their usual meanings. For 100 mm [Na⁺]_o, 100 mm [Li⁺]_o and 150 mm [K⁺]_i, these expressions give: $P_{\rm Na}$: $P_{\rm K}$: $P_{\rm Li}$ = 1:0·35:0·55. It should be mentioned, however, that variations in reversal potential observed in response to changes in external Na⁺ and Li⁺ concentrations were not in a good agreement with theoretical predictions of the Goldman–Hodgkin–Katz formula. One reason for this may be the deviation of the internal K⁺ concentration from 150 mm and its variation for different cells. However, we do not exclude the possibility of a concentration dependence of relative permeabilities for modified LTI Ca²⁺ channels.

2

The activity of modified Ca²⁺ channels was preserved after excision of the membrane path and formation of the inside-out configuration (Fig. 6). However, the mean current from excised patches had a smaller amplitude due to both a decrease in total channel open time during depolarization and an increase in the number of blank records (Fig. 6). Such reduced activity could be observed for a long time with no sign of run-down.

Ca²⁺ channel activation kinetics

A four-state kinetic model for the description of Ca²⁺ channel activation has recently been proposed (Kostyuk *et al.* 1989b):

$$R \stackrel{2\alpha}{\underset{\beta}{\leftarrow}} C \stackrel{\alpha}{\underset{2\beta}{\leftarrow}} A \stackrel{\alpha}{\underset{b}{\leftarrow}} O, \tag{1}$$

where R, C and A are closed states, O is the open state of the channel and α , β , a and b are the rate constants of channel transitions between the states. In terms of this model an adequate description of all experimental histograms was possible. Formally, the first two stages in model (1) can be attributed to a potential-dependent transfer of two charged gating particles in conformity with m^2 activation kinetics (Kostyuk, Krishtal & Shakhovalov, 1977), while the last kinetic stage describes subsequent fast (perhaps conformational) transformation of the channel gating mechanism which leads directly to channel opening.

The kinetics of Ca^{2+} channels in the neuroblastoma cells under investigation seem to be similar to those of LTI Ca^{2+} channels in neuronal membrane, the activation of which at the whole-cell level can be described using the m^2 Hodgkin–Huxley formalism (Fedulova *et al.* 1985). Thus, it was interesting to test the validity of model (1) for the present case and to examine how different kinetic stages of the channel activation may depend on permeant ion species. Although the model does not consider inactivation, we note that Carbone & Lux (1987 b) have suggested that Ca^{2+} channels can enter the inactive state from any foregoing state, and we discuss below the possible contribution of this to the measured rate constants.

The validity of model (1) was tested using the data-fitting procedure of Kostyuk et al. (1989b). The open time histogram could be adequately fitted with a single exponential (Fig. 7Aa and Ba) with time constant, $\tau_{\rm op}$, which is the inverse of rate constant b ($\tau_{\rm op} = 1/b$). The histogram of channel lifetime in the closed state required at least three exponentials to describe the data (Fig. 7Ab and Bb). The 'fast' exponential component of this histogram contained the highest fraction of all channel closures, and its time constant, $\tau_{\rm cl(f)}$, characterizes the mean value of intraburst shut times. In the framework of the model (1) this time constant can be expressed as (Shuba & Teslenko, 1987):

$$1/\tau_{\text{cl(f)}} \simeq a + 2\beta(1 + \alpha/a) \simeq a + 2\beta. \tag{2}$$

The burst duration histogram could be well approximated by the sum of two exponentials (Fig. 7Ac and Bc). The time constant of the 'slower' exponential,

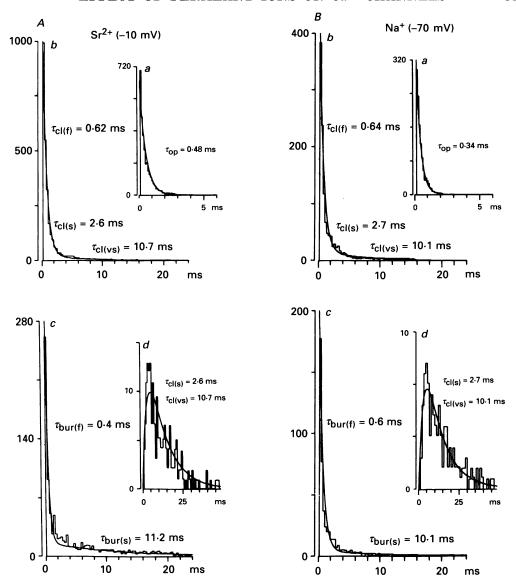


Fig. 7. Fitting of experimental time histograms for single Ca^{2+} channel activity in neuroblastoma cell membrane in conformity to model (1) (see text). Examples of open time (a), closed time (b), burst duration (c) and latency to first opening (d) histograms for 60 mm- Sr^{2+} (A) and 60 mm- Na^+ (B) as charge carriers. Smooth curves are single-(a), three-(b) and double-(c) and (a) exponential fittings (b) and (a), (b) difference of exponentials) to the experimental histograms. Corresponding time constants are shown near each histogram.

 $\tau_{\text{bur(s)}}$, is related to the rate constants a, b and β according to the following expression (Shuba & Teslenko, 1987; Kostyuk *et al.* 1989 b):

$$1/\tau_{\rm bur(s)} \simeq 2b\beta/(a+b). \tag{3}$$

Using eqns (2) and (3) and considering that $b = 1/\tau_{op}$ one can easily obtain the

relations which allow to calculate the rate constants a and β based on experimentally measured $\tau_{\rm op}$, $\tau_{\rm el(f)}$ and $\tau_{\rm bur(s)}$:

$$\begin{split} a &= (\tau_{\text{bur(s)}} - \tau_{\text{cl(f)}}) / \tau_{\text{cl(f)}} (\tau_{\text{bur(s)}} + \tau_{\text{op}}), \\ \beta &= (\tau_{\text{op}} + \tau_{\text{cl(f)}}) / 2\tau_{\text{cl(f)}} (\tau_{\text{bur(s)}} + \tau_{\text{op}}). \end{split}$$

The histogram of the first latencies to channel opening after depolarization onset could be satisfactorily fitted by the difference of two exponentials (Fig. 7Ad and Bd)

Table 1. Mean values of inverse rate constants a^{-1} , b^{-1} , α^{-1} , β^{-1} and time constant $\tau_{\text{cl(s)}}$ and $\tau_{\text{cl(vs)}}$ (in ms) for different permeant ions (all in 60 mm concentration) and different membrane potentials

Ion	$V_{\rm m}$ (mV)	a^{-1}	b^{-1}	α^{-1}	$oldsymbol{eta}^{-1}$	$ au_{ m cl(s)}$	$ au_{ m cl(vs)}$
Ca^{2+}	-30	0.53	0.56	20.3	9.6	4.0	51.6
	-20	0.58	0.54	16.2	10.8	3.8	33.7
	-10	0.50	0.40	9.5	12.5	3.0	15.1
	0	0.44	0.36	7.5	13.8	2.7	10.6
Sr^{2+}	-30	0.54	0.59	16.4	8.6	3.5	39.1
	-20	0.46	0.43	11.7	11.0	3.3	21.0
	-10	0.42	0.36	7.4	13·1	2.6	10.1
	0	0.34	0.29	5.4	16.5	$2\cdot 2$	6.8
$\mathbf{Ba^{2+}}$	-30	0.77	0.62	30.3	6.6	3.7	123.7
	-20	0.53	0.50	18.7	7.6	3.6	49.1
	-10	0.46	0.45	10.8	10.3	3.0	19.2
	0	0.40	0.36	5 ·8	14.5	$2 \cdot 2$	7.6
Na^+	-90	0.65	0.40	17.0	8.6	3.7	39.0
	-80	0.67	0.39	11.5	10.7	3.3	20.1
	-70	0.61	0.39	7.4	14.8	2.7	10.1
	-60	0.57	0.37	4·1	19.2	1.8	4.8

having the same time constants as 'slow' $(\tau_{\rm cl(s)})$ and 'very slow' $(\tau_{\rm cl(vs)})$ exponential components of the overall closed time histograms. These time constants allow one to determine the rate constant α using the expression (Kostyuk *et al.* 1989 *b*):

$$\alpha = (2\tau_{\rm cl(s)}\tau_{\rm cl(vs)})^{-0.5}.$$

It is necessary to point out that the described procedure of data processing was valid for an adequate description of channel activation irrespective of permeant ion species (see Fig. 7). This allowed us to determine a full set of rate constants describing channel activation in the framework of model (1) for different charge carriers (see Table 1). Figure 8A and B shows the potential dependence of mean channel open time and mean time of intraburst channel closures for divalent and monovalent charge carriers. The absolute values of these time constants and their potential dependence for $\operatorname{Ca^{2+}}$, $\operatorname{Sr^{2+}}$ and $\operatorname{Ba^{2+}}$ are very similar. Modification of normal channel selectivity that allowed high permeability to monovalent cations led first of all to a hyperpolarizing shift of $\tau_{\operatorname{op}}(V)$ and $\tau_{\operatorname{cl(f)}}(V)$ compared to divalent charge carriers. The absolute values of $\tau_{\operatorname{cl(f)}}$ for $\operatorname{Na^{+}}$ and $\operatorname{Li^{+}}$ were almost the same as for $\operatorname{Ca^{2+}}$, $\operatorname{Sr^{2+}}$ and $\operatorname{Ba^{2+}}$ and varied only slightly with changes in membrane potential. However, τ_{op} showed more prominent variations which depended on both permeant monovalent

ion type and concentration. The potential dependence of the mean channel open time for Na⁺ was much weaker, and absolute values smaller, than for Li⁺ and divalent charge carriers. Values for τ_{op} also depended on Na⁺ concentration, being smaller for 100 mm [Na⁺]_o than for 40 mm [Na⁺]_o (Fig. 8A).

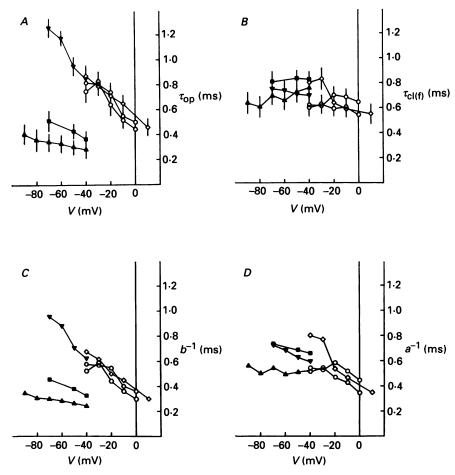


Fig. 8. Potential dependence of mean channel open time $\tau_{\rm op}$ (A) and mean time of intraburst closures $\tau_{\rm cl(f)}$ (B) for various permeant cations: \bigcirc , ${\rm Ca^{2+}}$; \bigcirc , ${\rm Sr^{2+}}$; \diamondsuit , ${\rm Ba^{2+}}$; \blacktriangle , 100 mm-Na⁺; \blacksquare , 40 mm-Na⁺; \blacktriangledown , 100 mm-Li⁺. C and D, calculated values of inverse rate constants b^{-1} (C) and a^{-1} (D) for the same membrane potentials and charge carriers. Calculations were made after correction of $\tau_{\rm op}$ and $\tau_{\rm cl(f)}$ for missed events according to expressions (4) and (5) (see text). Each symbol indicates experimental values for at least three patches. Vertical bars, s.D. from the mean. All divalent charge carriers were 60 mm.

Since the time constants $\tau_{\rm op}$ and $\tau_{\rm cl(f)}$ are of relatively short duration missed events due to frequency limitations of the recording apparatus (2 kHz for Na⁺-carried currents and 1 kHz for all others) could affect their numerical values. Using the results of Blatz & Magleby (1986) it is possible to show that if $a, b \gg \alpha$, β and $t_{\rm d} \ll \tau_{\rm op}$, $\tau_{\rm cl(f)}$, (which are true for Ca²⁺ channels and our recording conditions), the

real channel open time $T_{\rm op}$ and the real time of intraburst gaps $T_{\rm cl(f)}$ are related to the corresponding time $\tau_{\rm op}$ and $\tau_{\rm cl(f)}$ measured at a given bandwidth as follows:

$$1/b = T_{\rm op} \simeq \tau_{\rm op} (1 - t_{\rm d}/\tau_{\rm cl(f)}),$$
 (4)

$$1/(a+2\beta) = T_{\rm cl(f)} \simeq \tau_{\rm cl(f)} (1 - t_{\rm d}/\tau_{\rm op}).$$
 (5)

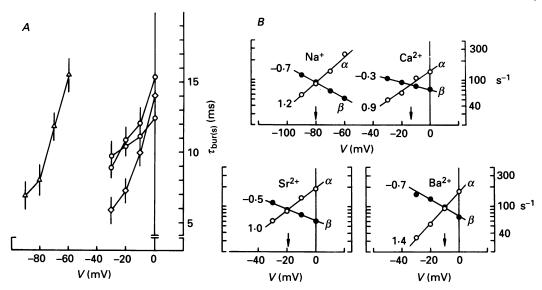


Fig. 9. Potential dependence of burst duration $\tau_{\text{bur(s)}}(A)$ and of rate constants α and β (B) for different charge carriers: \bigcirc , Ca^{2+} ; \bigcirc , Sr^{2+} ; \bigcirc , Ba^{2+} ; \bigcirc , Na^+ (all 60 mm). Straight lines in B are the best fit by eye to the experimental points. Corresponding semilogarithmic slopes are shown near each line. Arrows in B indicate intersection potentials of α - and β -lines.

The results of the definition of inverse rate constants b^{-1} and a^{-1} using corrected values of $T_{\rm op}$ and $T_{\rm cl(f)}$ (below we shall use usual denotations, $\tau_{\rm op}$ and $\tau_{\rm cl(f)}$, implying corrected values) are given in Fig. 8C and D. One can note stronger dependence of b^{-1} compared to a^{-1} upon permeant ion species and concentration.

Burst duration for all charge carriers was found to be strongly potential dependent, increasing with depolarization. However, the steepness of $\tau_{\text{bur(s)}}(V)$ for various ions was not similar; it was weakest for Ca^{2+} and strongest for Ba^{2+} and Na^+ (Fig. 9A). Figure 9B presents semilogarithmic plots of rate constants α and β . These rate constants define the character of activation curves for corresponding whole-cell currents. Experimental points could be satisfactory fitted with the straight lines indicating exponential dependence of rate constants α and β on membrane potential. The intersection points between the α - and β -lines correspond to the membrane potentials of half-activation of the channel. The observed shifts of $V_{\frac{1}{2}}$ in response to substitution of charge carriers are in good agreement with those obtained from I-V relations for whole-cell currents (Fig. 1) and activation curves (Fig. 3C).

Semilogarithmic slopes of α and β for various charge carriers are different. As the initial two kinetic stages in model (1) can formally be related to the intramembrane

transfer of two charged gating particles, the slope of $\ln \alpha(V)/\beta(V)$ represents the effective valency of each m-particle (m is an activation variable). Hence it is possible to conclude that this value is noticeably affected by permeant ion species.

DISCUSSION

Identification of Ca²⁺ channel type

The data obtained confirm our suggestion that the Ca²⁺ channels in undifferentiated neuroblastoma cells most resemble the LTI (T-type) Ca²⁺ channels in other preparations. These channels are similar in their macroscopic kinetics and selectivity properties and are practically independent of the intercellular metabolic support. Shifts of Ca²⁺ channel potential-dependent characteristics in neuroblastoma cell membrane caused by substitution of permeant divalent cations are also quite similar to those for LTI channels (e.g. Fukushima & Hagiwara, 1985).

Ca^{2+} channel selectivity

Ca²⁺ channel permeability is known to be determined by the depth of energy wells representing ion binding sites (for review see Tsien, Hess, McCleskey & Rosenberg, 1987) and by the height of the potential barrier which in general depends on electrostatic interaction of the permeant ion with the anionic groups of the selectivity filter, and sterical correspondence of hydrated ion to its pore (Hille, 1975). Since LTI Ca²⁺ channel conductance in our experiments was found to be identical for Ca²⁺, Sr²⁺ and Ba²⁺ it is possible to suppose that the affinities of these ions to channel binding sites and the height of the potential barriers for them are also quite similar.

Permeability of the channel for monovalent cations depends mainly upon the sterical factors. Recently, based on the fact that the largest organic cation able to permeate through the modified HTN (or L-type) Ca2+ channel was tetramethylammonium (TMA), a circle of 0.6 nm in diameter was proposed for its sterical part (McCleskey & Almers, 1985; Tsien et al. 1987). However, noticeable permeability of TMA through the LTI Ca²⁺ channels in lymphocytes has not been detected (Fukushima & Hagiwara, 1985). We are inclined to the idea that the dimensions of the sterical regions for Ca²⁺ channels of different types are similar and are close to those for Na⁺ channel (0·3 × 0·5 nm (Hille, 1975)). Indeed, all Ca²⁺ channels after modification of their selectivity can pass almost the same monovalent cations as Na+ channels (Kostyuk et al. 1983; Fukushima & Hagiwara, 1985; McCleskey & Almers, 1985). Some peculiarities of the selectivity properties may be explained by different flexibility of sterical regions. If so, then for some channels the size of the pore of the selectivity filter may not exactly correspond to the diameter of the largest permeable ion. Additional arguments in favour of this idea come from structural homology of Na⁺ and Ca²⁺ channels (Miller, 1989).

The effect of permeant ions on mean channel open time

The data obtained show that permeant ion type and concentration can noticeably affect the mean Ca^{2+} channel open time, τ_{op} , which characterizes the rate of Ca^{2+} channel transition from open (O) into nearest closed state (A). This suggests that channel closure is probably due to sterical narrowing of the selectivity filter.

Recently, it was shown by Läuger (1985) that ions which occupy channel binding site(s) can affect the rate constants of conformation transitions accompanying channel gating due to electrostatic interaction with the polar groups of channel protein. Although we consider such an 'electrostatic mechanism' important, the difference in the ion dependence of the inverse rate constants a^{-1} and b^{-1} cannot be explained by this mechanism alone. In our opinion to account for the stronger ion dependence of b^{-1} which characterizes the mean channel open time not only static, but also dynamic interaction of the permeating ion with polar molecular groups of Ca^{2+} channel selectivity filter should be considered.

Such interaction may occur if the hydrated ion passing through the open Ca^{2+} channel does not sterically correspond to the cross-section of the selectivity filter. In this case the local displacements, δ_i , of charged molecular groups and consequently the system of dipoles, $e\delta_i$ ($e=1.6\times10^{-19}\,\mathrm{C}$), may occur along the ion trajectory. The mean induced dipole moment will depend on the relation between the ion permeation rate and the rate of dipole relaxation, which for proteins is of the order of $10^7\,\mathrm{s}^{-1}$ (Burfoot & Taylor, 1979). According to Debye's theory the mean dipole moment determines the effective dielectric constant for a system of weakly interacting molecular groups (Burfoot & Taylor, 1979). Thus, in the presence of ionic flux one might expect increased dielectric constant, ϵ_i , of the channel sterical region compared to the value $\epsilon_0 \simeq 3$ which is characteristic for static proteins (Honig, Hubbel & Flewelling, 1986). This, in turn, will result in a decrease in part of the transmembrane electric field falling on this region and governing the stage of channel conformational closure. Such a process may be detected as a decrease in the potential dependence of the mean channel open time.

Since the real dimensions of the pore of the LTI Ca²⁺ channel selectivity filter (as well as of other channels) are unknown it is difficult to make quantitative estimates related to this mechanism. However, we have found that at relatively high Na⁺ concentrations when $g_{\mathrm{Na}} \simeq 60~\mathrm{pS}$ the rate of Na⁺ ion transfer through the channel starts to approach the rate of dipole relaxation. For such conditions it is sufficient to suppose a local displacement, δ_{Na} , of only 0.016 nm to obtain the increase in the dielectric constant of the channel sterical region from ϵ_0 up to $\epsilon_{\rm Na} \simeq 30$. Divalent charge carriers and Na+ ions at low [Na+]o due to the decrease of ion permeation rate are able to increase the dielectric constant of the sterical region by about 10% at maximum compared to its static value. Finally, Li⁺ ions, as the smallest, may not induce any dipole moment ($\delta_{Li} = 0$) and the dielectric constant of the sterical region in the presence of Li⁺ flux will not increase at all $(\epsilon_{Li} = \epsilon_0)$. Figure 8 C shows that the potential dependence of τ_{op} ($\tau_{op} = b^{-1}$) reasonably reflects expected changes of the dielectrical constant produced by different ions. This potential dependence is highest for Li⁺, decreases for divalent charge carriers and Na⁺ at low concentrations, and almost disappears for high $[\mathrm{Na^+}]_{\mathrm{o}}$; in parallel shortening of τ_{op} occurs.

The proposed mechanism for the ion dependence of $\tau_{\rm op}$ is consistent with previous observations on Ca²⁺ channels from rat brain synaptosomes (Nelson, 1986). For these Ca²⁺ channels (probably high threshold), the mean open time was found to be shortest for Ba²⁺ ions which had the highest permeation rate (and biggest ionic radius), and it increased in parallel with the decrease of single-channel conductance (and ionic radius) for Sr²⁺, Ca²⁺ and Mn⁺.

When studying the activation kinetics of single LTI Ca²⁺ channels in monovalent ion conducting mode we kept the external Ca²⁺ concentration at 10^{-8} M which is low enough to exclude specific blocking action of Ca²⁺ ions. We expect that increasing $[Ca^{2+}]_o$ will lead to the appearance of blocking events resulting in the reduction of mean channel open time as was found by Lux, Carbone & Zucker (1988) on LTI Ca²⁺ channels from chick DRG neurons. In this preparation τ_{op} for sodium current was halved at $[Ca^{2+}]_o \simeq 5 \times 10^{-5}$ M.

The effect of permeant ions on the channel closed states

After channel transition from the open state (O) to the nearest closed state (A) the sterical region becomes narrow and ion flux through the channel is no longer possible. That is why the inverse rate constant a^{-1} which characterizes the backward transition to an open state is less ion-dependent than b^{-1} .

Variation in the permeant ion type causes changes in the logarithmic slopes of $\alpha(V)$ and $\beta(V)$ and shifts of the half-activation potential of the channel. These effects can be explained by supposing that $\operatorname{Ca^{2+}}$ channel binding sites have a geometrical arrangement along the pathway of gating charge transfer. In this case, consistent with the 'electrostatic mechanism' proposed by Läuger (1985), the energies of the states R and C will include the component describing electrostatic interaction between the bound cation and the gating charge. A high-affinity binding site in the $\operatorname{Ca^{2+}}$ channel outer mouth responsible for selectivity modification (Kostyuk et al. 1983) can add much to such interaction if the initial localization of charged gating particles in the channel resting state (R) is in a close proximity to this binding site. The 70 mV hyperpolarizing shift of the activation curve for the modified $\operatorname{Ca^{2+}}$ channel, which is much larger than the possible shift due to an increase in surface potential in the absence of divalent cations ($V_{\rm s}=-30~{\rm mV}$; Mironov & Dolgaya, 1987), indicates that this might be the case.

Contribution of inactivation

The Ca²⁺ channels in neuroblastoma cells as well as LTI Ca²⁺ channels in other preparations are relatively fast-inactivating in a voltage-dependent manner (Bean, 1985; Fedulova *et al.* 1985; Carbone & Lux, 1987 a). According to our data, the time constant ($\tau_{\rm in}$) of Sr²⁺ current decay during sustained depolarization to 0 mV is 17 ms (not shown). Since theoretically the LTI Ca²⁺ channel can enter the inactive state (I) from each state of the activation process (Carbone & Lux, 1987 b; Droogmans & Nilius, 1989) all rate constants measured in the present investigation can be influenced by inactivation and this influence needs to be assessed.

First, inactivation results in the appearance of additional exponential components in the closed time histograms having time constants bigger than $\tau_{\rm cl(f)}$, $\tau_{\rm cl(s)}$ and $\tau_{\rm cl(vs)}$. In fact, we observed some 'extra slow tails' in these histograms. However, due to the relative infrequency of events contributing to these 'tails' and due to their contamination by false closings (see Methods), an adequate description was impossible. In determining the rate constant α , first latency histograms were used and these are minimally affected by inactivation. In contrast, mean burst duration can be strongly affected by inactivation due to channel transition from state A into inactive state I. Thus, the apparent difference in potential dependence of $\tau_{\rm bur(s)}$ and

slopes of β (Fig. 9) can partially be attributed to the effect of various ions on channel inactivation. Indeed, some difference in inactivation time course of Ca²⁺, Sr²⁺ and Ba²⁺ whole-cell currents can be noted in Fig. 1A. Slowing of inactivation of the LTI Ba²⁺ current compared to the Ca²⁺ current in the DRG neurons was also detected by Carbone & Lux (1987 a).

Consideration of the additional transition of the channel from open state O into inactive state I due to the big difference between $\tau_{\rm op}$ and $\tau_{\rm in}$ may change the rate constant b defining channel conformational closure by not more than 2%. Thus, this rate constant can be considered inversely proportional to $\tau_{\rm op}$ without introducing a significant error.

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